

# Dynamic Properties of the G93A Mutant of Copper–Zinc Superoxide Dismutase As Detected by NMR Spectroscopy: Implications for the Pathology of Familial Amyotrophic Lateral Sclerosis<sup>†</sup>

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**ABSTRACT:** The backbone assignment of the copper–zinc superoxide dismutase amyotrophic lateral sclerosis G93A mutant was performed on an <sup>15</sup>N-enriched protein sample. <sup>15</sup>N *R*<sub>1</sub>, *R*<sub>2</sub>, and *R*<sub>1ρ</sub> and <sup>15</sup>N–<sup>1</sup>H NOE experiments were then carried out at 600 MHz on G93A Cu<sub>2</sub>Zn<sub>2</sub>SOD and the values compared to the dynamics data for the “wild-type” protein. In addition, <sup>15</sup>N and <sup>1</sup>H chemical shift comparisons between wild-type Cu<sub>2</sub>Zn<sub>2</sub>SOD and its G93A mutant were also made. G93A exhibits a higher mobility than wild-type Cu<sub>2</sub>Zn<sub>2</sub>SOD, particularly in loops III and V, on a time scale faster than the rate of protein tumbling. There are also distinct chemical shift and NOE differences in residues 35–42 and 92–95, which comprise these loops. These two regions of Cu<sub>2</sub>Zn<sub>2</sub>SOD form the end of the β-barrel termed the “β-barrel plug” [Tainer, J. A., Getzoff, E. D., Beem, K. M., Richardson, J. S., and Richardson, D. C. (1982) *J. Mol. Biol.* 160, 181–217]. The increased mobility and reduction of the number of observed NOEs in this region indicate an opening of the β-barrel that may lead to amyloid fibrillogenesis. Alternatively, a motor neuron-specific substrate may bind this region of the protein, leading to deleterious modifications and/or reactions.

Amyotrophic lateral sclerosis (ALS)<sup>1</sup> is a progressive neurodegenerative disease that results in a gradual degradation of motor neurons in the motor cortex, brain stem, and spinal cord (1). A genetic link has been found in roughly 10% of all reported cases of the disease, which are collectively termed familial ALS (FALS). Of these cases, approximately one in five has been directly attributed to autosomal dominant mutations in the gene that encodes copper–zinc superoxide dismutase (Cu<sub>2</sub>Zn<sub>2</sub>SOD) (2–5). An intriguing aspect of these mutations is that they occur throughout the gene and yet in most instances do not significantly lower the SOD activity of the resulting enzyme (6).

At present, the mechanism or mechanisms whereby mutations in the gene encoding Cu<sub>2</sub>Zn<sub>2</sub>SOD cause the disease

are unknown. A major insight into the role of Cu<sub>2</sub>Zn<sub>2</sub>SOD in FALS was gained when Gurney et al. found that transgenic mice expressing the FALS mutant G93A Cu<sub>2</sub>Zn<sub>2</sub>SOD developed progressive motor neuron disease despite having high levels of SOD activity (7). In addition, transgenic mice with no endogenous Cu<sub>2</sub>Zn<sub>2</sub>SOD were found not to develop motor neuron disease, suggesting that SOD-associated FALS is caused by a toxic gain of function induced by the ALS mutant Cu<sub>2</sub>Zn<sub>2</sub>SOD protein (8).

Cu<sub>2</sub>Zn<sub>2</sub>SOD is a 32 kDa homodimer containing two copper and two zinc ions per dimer (9). Reduction and oxidation of the copper ion mediates the disproportionation of superoxide anion to hydrogen peroxide and dioxygen ( $2\text{O}_2^- + 2\text{H}^+ \rightarrow \text{O}_2 + \text{H}_2\text{O}_2$ ) (10). The tertiary structure of the enzyme is a Greek key β-barrel containing an internal disulfide bond (11, 12), both of which contribute to its high stability, as characterized by its resistance to thermal denaturation as well as to proteolysis (13). It has been postulated that structural destabilization and/or an increase in mobility of Cu<sub>2</sub>Zn<sub>2</sub>SOD mutants may play a role in the pathology of Cu<sub>2</sub>Zn<sub>2</sub>SOD-related ALS (14–17). Conserved glycine residues are common in almost all of the known Cu<sub>2</sub>Zn<sub>2</sub>SOD proteins (18). Moreover, five missense mutations in the conserved glycine at residue 93 have been implicated in SOD-related FALS (3–6, 19).

Numerous attempts were made to express and purify the most prominent Cu<sub>2</sub>Zn<sub>2</sub>SOD FALS mutant, A4V, prior to purifying G93A (unpublished work). The yield of the former mutant was much lower than for G93A. The A4V mutant experienced degradation, as found from <sup>15</sup>N-HSQC spectra

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<sup>1</sup> Abbreviations: ALS, amyotrophic lateral sclerosis; FALS, familial amyotrophic lateral sclerosis; Cu<sub>2</sub>Zn<sub>2</sub>SOD, copper–zinc superoxide dismutase; NOE, nuclear Overhauser effect; HSQC, heteronuclear single-quantum coherence; NOESY, nuclear Overhauser effect spectroscopy; CPMG, Carr–Purcell–Meiboom–Gill; *R*<sub>1</sub>, longitudinal relaxation rate; *R*<sub>2</sub>, transverse relaxation rate.

just 2–3 days after purification. These results are indicative of the inherent instability of the Cu<sub>2</sub>Zn<sub>2</sub>SOD FALS mutants at room temperature. Fortunately, G93A was stable enough that all the needed experiments in this study could be performed.

In the study presented here, the dynamic properties of the most prevalent of the G93X FALS mutants, G93A, are investigated by nuclear magnetic resonance spectroscopy and compared to those of the wild-type protein. As the aim of this study is to characterize the spectral and dynamical differences between the G93A mutant and wild-type Cu<sub>2</sub>Zn<sub>2</sub>SOD, we have performed the measurements with the same setup that was used for the already reported wild-type characterization (17). This is the first study of the dynamic properties of a Cu<sub>2</sub>Zn<sub>2</sub>SOD-related FALS mutant in which variations in mobility are characterized in distinct residues of the protein.

## EXPERIMENTAL PROCEDURES

**Sample Preparation.** The human G93A Cu<sub>2</sub>Zn<sub>2</sub>SOD plasmid construct was created by PCR mutagenesis using the pCuZnSODII<sup>9</sup> plasmid (20) as a template. The template plasmid also contains the C6A and C111S mutations, which increase the thermostability of the sample and do not have discernible effects on the activity or the structure of Cu<sub>2</sub>Zn<sub>2</sub>SOD (13, 21) (recombinant Cu<sub>2</sub>Zn<sub>2</sub>SOD with these two mutations is termed “wild type” in this paper). This is the same plasmid that was used for the earlier dynamics studies of wild-type Cu<sub>2</sub>Zn<sub>2</sub>SOD (17). After PCR mutagenesis, the resulting plasmid was sequenced and then expressed in the *Escherichia coli* TOPP1 (Stratagene) strain (22) to obtain the uniformly <sup>15</sup>N-labeled dimeric protein. The <sup>15</sup>N-enriched sample, termed G93A in this paper, was then isolated and purified according to a published procedure (20).

Protein purity was verified by ESI-mass spectrometry. The G93A Cu<sub>2</sub>Zn<sub>2</sub>SOD mutant was found to have 2.03 Cu and 2.3 Zn equivalents per dimer by inductively coupled plasma emission spectroscopy, which is consistent with a Cu<sub>2</sub>Zn<sub>2</sub>SOD holoenzyme. Differential scanning calorimetry was also performed as previously described to ensure sample homogeneity (15). Protein concentrations were determined by UV spectroscopy using a molar extinction coefficient of  $1.08 \times 10^4 \text{ M}^{-1} \text{ cm}^{-1}$  at 280 nm for the purified enzyme. The final protein sample, at a concentration of 2.5 mM in 20 mM phosphate (pH 5.0), was reduced with 5–7 mM isoascorbate and loaded into an NMR tube in a VacAtmosphere chamber under nitrogen at 12 °C.

**NMR Spectroscopy.** The NMR experiments for the backbone assignment were carried out at 298 K on a Bruker Avance spectrometer operating at 700 MHz. Two-dimensional (2D) <sup>15</sup>N-HSQC, 2D NOESY, three-dimensional (3D) <sup>15</sup>N-NOESY-HSQC, and HNHA experiments were performed to obtain the backbone assignments. The two-dimensional NOESY spectrum was recorded with a spectral width of (<sup>1</sup>H) 18 250 Hz × (<sup>1</sup>H) 18 250 Hz, using (<sup>1</sup>H) 2048 × (<sup>1</sup>H) 896 data points. The three-dimensional <sup>15</sup>N-NOESY-HSQC and HNHA spectra were both collected with (<sup>1</sup>H) 10 504 Hz × (<sup>15</sup>N) 2980 Hz × (<sup>1</sup>H) 10 504 Hz as the spectral window. There were (<sup>1</sup>H) 1024 × (<sup>15</sup>N) 56 × (<sup>1</sup>H) 400 and (<sup>1</sup>H) 1024 × (<sup>15</sup>N) 36 × (<sup>1</sup>H) 200 data points, respectively. All mobility NMR experiments were carried out at 298 K

on a Bruker Avance spectrometer operating at 600 MHz. This spectrometer contained a triple-resonance (TXI) 5 mm probe equipped with pulsed field gradients (PFG) along the *z*-axis. In this study, all the relaxation measurements were collected with the same procedures used for the wild-type Cu<sub>2</sub>Zn<sub>2</sub>SOD (17) to ensure meaningful comparison between the two sets of data and avoid systematic errors. *R*<sub>1</sub>, *R*<sub>2</sub>, and heteronuclear NOEs were measured with the gradient-enhanced, sensitivity-enhanced pulse sequences as described by Farrow et al. (23). Constant temperature throughout the acquisition of data was maintained by the application of temperature compensation methods (24); possible shift variations were minimized by interleaving experiments with different delays (25). The <sup>15</sup>N longitudinal relaxation rates (*R*<sub>1</sub>) were measured with delays of 2.5, 35, 75, 270, 400, 650, 750, 1000, 1750, 2500, and 3000 ms. The <sup>15</sup>N transverse relaxation rates (*R*<sub>2</sub>) were measured using a CPMG sequence (26) with delays of 9.8, 19.5, 29.3, 39.0, 48.4, 58.6, 68.3, 78.1, 87.8, 97.7, 107.4, and 117.1 ms and a refocusing time (*τ*<sub>CPMG</sub>) of 225 μs. In both *R*<sub>1</sub> and *R*<sub>2</sub> measurements, the first delay was duplicated. A recycle delay of 3 s was used for *R*<sub>1</sub>, *R*<sub>2</sub>, and heteronuclear NOEs without proton saturation measurements. In the heteronuclear NOE experiment with proton saturation, the latter was achieved with a train of 120° <sup>1</sup>H pulses at 5 ms intervals applied for 3 s. All experiments utilized the “water flipback” suppression scheme to minimize the saturation of water by orienting it along the +*z*-axis during acquisition (27). All 2D data consisted of 2K data points in the acquisition dimension and of 256 experiments in the indirect dimension with eight scans and spectral widths of 8928 (<sup>1</sup>H) and 2553 Hz (<sup>15</sup>N).

To monitor contributions to transverse relaxation from exchange processes, the *R*<sub>2</sub> values of the <sup>15</sup>N nuclei were also measured as a function of *τ*<sub>CPMG</sub> (26, 28). Sets of experiments were carried out at six *τ*<sub>CPMG</sub> values: 225, 350, 450, 575, 700, and 1000 μs. Relaxation delays varied from 9 to 120 ms, with the exact values depending on *τ*<sub>CPMG</sub>.

The <sup>15</sup>N off-resonance rotating-frame relaxation rates (*R*<sub>1ρ</sub><sup>OFF</sup>) were measured as a function of the effective magnetic field amplitude (*ω*<sub>eff</sub>) using a previously reported pulse sequence (29). *ω*<sub>eff</sub> was modulated by changing both the *ω*<sub>1</sub> frequency and amplitude which has an offset (*Δω*) with respect to the center of the amide nitrogen resonances. This determines the effective magnetic field amplitude *ω*<sub>eff</sub> [ $= (\Delta\omega^2 + \omega_1^2)^{1/2}$ ], which makes an angle of *θ* [ $= \arctan(\omega_1/\Delta\omega)$ ] with the *y*-axis of the <sup>15</sup>N magnetization. The *ω*<sub>1</sub> amplitude was increased and decreased gradually in a trapezoidal fashion to achieve adiabatic rotation of the magnetization to the effective magnetic field axis (30). For each *ω*<sub>1</sub> amplitude value, a series of 2D experiments were performed in which the relaxation delays were set to range from 10 to 200 ms.

**Data Analysis.** Relaxation rates *R*<sub>1</sub>, *R*<sub>1ρ</sub><sup>OFF</sup>, and *R*<sub>2</sub> were determined by fitting the cross-peak intensities (*I*), measured as a function of the delay (*T*) within the pulse sequence, to a single-exponential decay curve using the Levenburg–Marquardt algorithm (31, 32) according to the following equation:

$$I(T) = I_0[\exp(-RT)] \quad (1)$$

where *I*<sub>0</sub> and *R* are the adjustable fitting parameters. The

use of a two-parameter fitting reflects the setup of the experiments (the properties of the pulse sequences) where the magnetization tends to 0 at infinite delay. Examples of  $^{15}\text{N}$   $R_1$  and  $R_2$  data recorded for G93A  $\text{Cu}_2\text{Zn}_2\text{SOD}$  are reported in Figure 1 of the Supporting Information. The errors on the relaxation parameters were estimated through a Monte Carlo approach by fitting 500 simulated data sets (33) generated from the optimized parameters with the addition of a Gaussian noise of standard deviation equal to either 5% of the largest peak intensity or the semidispersion of the intensities in replicated experiments.

The heteronuclear  $^1\text{H}$ – $^{15}\text{N}$  NOE values were measured by considering the ratio of the peak volumes in spectra recorded with and without  $^1\text{H}$  saturation. Two duplicate spectra were analyzed in a similar way to derive the uncertainty of the measurements.

**Rotating Frame Relaxation Rates.** Rate  $R_{1\rho}$  of a  $^{15}\text{N}$  spin, measured applying an off-resonance  $\omega_{\text{eff}}$  ( $R_{1\rho}^{\text{OFF}}$ ), is related to relaxation and exchange parameters by the following expression (34, 35):

$$R_{1\rho}^{\text{OFF}} = R_1 \cos^2 \theta + R_{1\rho}^{\text{ON},\infty} \sin^2 \theta + K \sin^2 \theta \frac{\tau_{\text{ex}}}{1 + \tau_{\text{ex}}^2 \omega_{\text{eff}}^2} \quad (2)$$

where  $R_1$  is the longitudinal relaxation rate,  $R_{1\rho}^{\text{ON},\infty}$  is the on-resonance rotating frame relaxation rate with an infinitely large effective field amplitude, and  $\tau_{\text{ex}}$  is the correlation time for the exchange process for the spin under observation.  $K$  is a constant equal to  $p_a p_b \delta\Omega^2$ , where  $p_a$  and  $p_b$  are the relative populations of the two states, a and b, between which the exchange process occurs and  $\delta\Omega$  is the chemical shift difference of the resonating nucleus between these two states. Equation 2 is valid when  $\delta\Omega\tau_{\text{ex}} \ll 1$  and off-resonance effects are neglected (36).

To compare the  $R_{1\rho}$  values measured with different  $\omega_{\text{eff}}$  values,  $R_{1\rho}^{\text{OFF}}$  should be corrected to take into account the different  $\theta$  angles (29):

$$R_{1\rho}^{\text{OFF,cor}} = \frac{R_{1\rho}^{\text{OFF}} - R_1 \cos^2 \theta}{\sin^2 \theta} = R_{1\rho}^{\text{ON},\infty} + K \frac{\tau_{\text{ex}}}{1 + \tau_{\text{ex}}^2 \omega_{\text{eff}}^2} \quad (3)$$

In the measurements presented here,  $\omega_{\text{eff}}$  varied between  $880 \times 2\pi$  and  $2200 \times 2\pi \text{ s}^{-1}$ , thus setting the range of the exchange process rates ( $\tau_{\text{ex}}^{-1}$ ) from  $7 \times 10^3$  to  $2 \times 10^4 \text{ s}^{-1}$ , the dependence of which can be detected in these experiments.

**CPMG Spin-Echo Pulse Train.** In the  $R_2$  measurements with the CPMG sequence, the contributions to transverse relaxation due to exchange processes are related to the  $\tau_{\text{CPMG}}$  length by the following equations:

$$R_{\text{ex}} = \frac{K_{\text{ex}}}{2} - 2\nu_{\text{eff}} \sinh^{-1} \left( \frac{K_{\text{ex}}}{\xi} \sinh \frac{\xi}{4\nu_{\text{eff}}} \right) \quad (4)$$

$$\nu_{\text{eff}} = \frac{1}{2(T_{\pi} + \tau_{\text{CPMG}})} \quad (5)$$

where  $\xi = (K_{\text{ex}}^2 - 4p_a p_b \delta\Omega^2)^{1/2}$ ,  $K_{\text{ex}} = 1/\tau_{\text{ex}}$ , and  $T_{\pi}$  is the duration of a  $180^\circ$   $^{15}\text{N}$  CPMG pulse. When  $\tau_{\text{ex}}$  is longer than  $\tau_{\text{CPMG}}$ , no effect due to conformational exchange processes is observed. Experimental limits on the spin-echo delay length are determined by the duty cycle of the transmitter for short delays and the evolution of  $^1\text{H}$ – $^{15}\text{N}$  coupling during long delays. In this study,  $\nu_{\text{eff}}$  ranges between 400 and  $1600 \text{ s}^{-1}$  which sets the limits for the detection of the frequency dependence of  $R_2$  to  $2.5 \times 10^3$  to  $1 \times 10^4 \text{ s}^{-1}$ .

**Model-Free Analysis.** The experimental longitudinal and transverse relaxation rates and the heteronuclear NOEs have been analyzed with the Model-Free approach, proposed by Lipari and Szabo (37), and its subsequently extended version proposed by Clore et al. (38).

The equations for the relaxation rates, in terms of spectral density functions  $[J(\omega)]$ , are well-known (39). Within the Model-Free analysis, the relaxation rates are expressed as a function of the overall rotational correlation time  $\tau_m$ , the generalized order parameter  $S^2$ , and the correlation time for internal motions, which can be considered to arise from two components, one describing faster ( $\tau_f$ ) and one slower ( $\tau_s$ ) motions (collectively called  $\tau_e$ ), but always faster than  $\tau_m$ . To ensure proper fitting of  $R_2$ , the Model-Free program also takes into account a contribution from exchange processes ( $R_{\text{ex}}$ ) when necessary.

The generalized order parameter  $S^2$  characterizes the amplitude of internal motions of the  $^1\text{H}$ – $^{15}\text{N}$  vector in each residue. A value of 1 indicates complete internal restriction of the  $^1\text{H}$ – $^{15}\text{N}$  bond, whereas a value of 0 represents no internal restriction.

Following a well-established procedure (38) initially, the relaxation rates are analyzed in terms of spectral densities, which, in the simplest case and for isotropic tumbling, is

$$J(\omega) = \frac{2}{5} \left[ \frac{S^2 \tau_m}{1 + (\omega \tau_m)^2} \right] \quad (6)$$

where the only variable parameter is  $S^2$ ,  $\tau_m$  being the overall rotational correlation time (model 1).

An  $F$ -statistic test (40) can be applied to evaluate the quality of the fitting with respect to the model used for data analysis. When the fitting using eq 6 is poor, a second term is added to the spectral density function, according to the following equation:

$$J(\omega) = \frac{2}{5} \left[ \frac{S^2 \tau_m}{1 + (\omega \tau_m)^2} + \frac{(1 - S^2) \tau'}{1 + (\omega \tau')^2} \right] \quad (7)$$

where  $\tau'^{-1} = \tau_m^{-1} + \tau_e^{-1}$  and  $\tau_e$  is the correlation time for internal motions faster than  $\tau_m$  (model 2).

When internal motions of significant amplitude occur on two different time scales, but still faster than  $\tau_m$ , the Model-Free spectral density function extends to

$$J(\omega) = \frac{2}{5} \left[ \frac{S^2 \tau_m}{1 + (\omega \tau_m)^2} + \frac{(S_f^2 - S^2) \tau'_s}{1 + (\omega \tau'_s)^2} + \frac{(1 - S_f^2) \tau'_f}{1 + (\omega \tau'_f)^2} \right] \quad (8)$$

where the slow and fast motions have different correlation times ( $\tau_s$  and  $\tau_f$ ),  $\tau'^{-1}_s = \tau_m^{-1} + \tau_s^{-1}$  and  $\tau'^{-1}_f = \tau_m^{-1} + \tau_f^{-1}$ ,  $S^2 (=S_f^2 S_s^2)$  is the square of the generalized order parameter characterizing the amplitude of the internal motions, and  $S_f^2$



and  $S_s^2$  are the squares of the order parameters for the internal motions on the fast and slow time scales, respectively.

Since the number of parameters usually exceeds the number of experimentally accessible data, it is commonly assumed that  $\tau_f < 20$  ps. The third term of eq 8 is then sufficiently small that it makes a negligible contribution to the relaxation parameters (38, 41) and simplifies eq 8 to

$$J(\omega) = \frac{2}{5} \left[ \frac{S^2 \tau_m}{1 + (\omega \tau_m)^2} + \frac{(S_f^2 - S^2) \tau_s'}{1 + (\omega \tau_s')^2} \right] \quad (9)$$

to fit the data (model 5).

When the overall motion is axially symmetric (42),  $\tau_m^{-1}$  assumes the following values:  $\tau_{m1}^{-1} = 6D_{\perp}$ ,  $\tau_{m2}^{-1} = 5D_{\perp} + D_{\parallel}$ ,  $\tau_{m3}^{-1} = 2D_{\perp} + 4D_{\parallel}$ . Consequently, the spectral densities of eqs 6–9 become a summation over three components of the diffusion tensor. The three terms of the summation are multiplied by  $A_1 [(3 \cos^2 \theta - 1)/4]$ ,  $A_2 (=3 \sin^2 \theta \cos^2 \theta)$ , and  $A_3 (=3/4 \sin^4 \theta)$ , respectively, where  $\theta$  is the angle between the H–N bond vector and the unique axis of the principal frame of the diffusion tensor (43).

The rotational diffusion tensor ( $D$ ) and the initial  $\tau_m$  values were estimated from the  $R_2/R_1$  ratio with the program Quadric Diffusion (43, 44). Once the best model for the molecular motions was selected on the basis of an  $F$ -statistic test (40) of the Model-Free analysis, the overall  $\tau_m$ , the  $D_{\parallel}/D_{\perp}$  ratio, and the internal motion parameters for each spin were optimized by fitting the experimental relaxation data ( $R_1$ ,  $R_2$ , and NOE) using the Powell minimization algorithm (35). In this analysis, an axially symmetric model was used to account for the global rotational motion of the protein.

**Estimation of Conformational Entropy from NMR Relaxation Data.** Order parameters ( $S^2$ ) are related to conformational entropy ( $S_{\text{conf}}$ ) because they indicate the range of structural states accessible to a bond vector (45). It has been previously shown that, in the assumption that the motions of the individual bond vectors are independent, an upper limit (in magnitude) of the entropy changes occurring between two states of a protein can be calculated from order parameters (46):

$$\Delta S/k_b = \ln \left[ \frac{3 - (1 + 8S_{n,2})^{1/2}}{3 - (1 + 8S_{n,1})^{1/2}} \right] \quad (10)$$

where  $S$  is the conformational entropy,  $k_b$  is Boltzmann's constant, and  $S_n$  are the order parameters of the  $n$ th H–N vectors in the two states. In this specific case, the initial state (state 1) is the wild-type Cu<sub>2</sub>Zn<sub>2</sub>SOD and the final state (state 2) is the G93A Cu<sub>2</sub>Zn<sub>2</sub>SOD mutant.

## RESULTS

**Resonance Assignment of G93A Cu<sub>2</sub>Zn<sub>2</sub>SOD.** The G93A Cu<sub>2</sub>Zn<sub>2</sub>SOD mutant exhibits a well-dispersed and well-resolved <sup>15</sup>N-HSQC spectrum indicative of a folded protein. However, this mutant, particularly at the concentrations needed for NMR characterization (around 1 mM), is unstable and after ~2 weeks at room temperature starts to denature and eventually becomes completely unfolded. This prevents us from performing a more extensive spectral characterization. Nevertheless, we have been able to assign its spectrum and to study its dynamical properties. The backbone assignment was aided by using the automatic assignment program

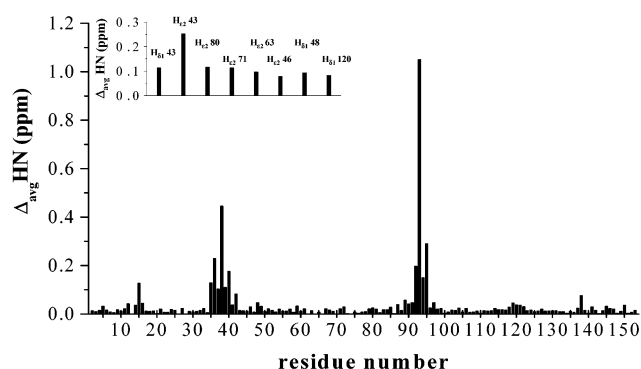


FIGURE 1: HN chemical shift differences between wild-type and G93A Cu<sub>2</sub>Zn<sub>2</sub>SOD. The weighted average differences ( $\Delta_{\text{avg}} \text{HN}$ ) in chemical shift are determined as  $\{[(\Delta H)^2 + (\Delta N/5)^2]/2\}^{1/2}$ , where  $\Delta H$  and  $\Delta N$  are chemical shift differences for <sup>1</sup>H and <sup>15</sup>N, respectively (61). The inset reports the  $\Delta_{\text{avg}} \text{HN}$  values for the HN groups of the histidine ring.

GARANT (47) run on a Linux processor cluster. The peak lists of <sup>15</sup>N-NOESY-HSQC and <sup>15</sup>N-HSQC experiments, as well as the solution structure of the Cu<sub>2</sub>Zn<sub>2</sub>SOD dimeric protein (48), were used as input for the program. Of the 147 expected cross-peaks in the <sup>15</sup>N-HSQC spectrum, only Phe64 and Lys70 could not be assigned. The <sup>15</sup>N-HSQC spectrum of G93A Cu<sub>2</sub>Zn<sub>2</sub>SOD was nearly identical to that of the wild-type protein (48), indicating structural similarity between the two, with the exception of a few areas of localized variations in chemical shift (Figure 1). These areas are very well-defined, being limited to segments of residues 14–16, 35–41, and 89–95. Furthermore, the NOE patterns observed from the <sup>15</sup>N-NOESY-HSQC spectra were virtually equal for all residues except for those experiencing variations in chemical shift with respect to the wild type. A number of residues located in the segments of residues 35–40 and 90–95 show a sizable reduction in the number and intensities of NOE cross-peaks.

**Relaxation Measurements.** The backbone NH dynamics of G93A Cu<sub>2</sub>Zn<sub>2</sub>SOD were investigated on both the millisecond to microsecond and nanosecond to picosecond time scales. The G93A Cu<sub>2</sub>Zn<sub>2</sub>SOD relaxation measurements were carried out under the same sample and experimental conditions that were used for the wild-type Cu<sub>2</sub>Zn<sub>2</sub>SOD (17) to avoid systematic errors between the two sets of data. We obtained reliable  $R_1$ ,  $R_2$ , and NOE data for 138 of the possible 145 residues that contain backbone NH groups. Glu40, Asn53, Ser59, Ser68, Val118, His120, and Asn131 could not be analyzed due to peak overlap or an inadequate peak signal/noise ratio.

The average  $R_1$ ,  $R_2$ , and NOE values for G93A Cu<sub>2</sub>Zn<sub>2</sub>SOD were  $0.62 \pm 0.06 \text{ s}^{-1}$ ,  $30.1 \pm 3.4 \text{ s}^{-1}$ , and  $0.83 \pm 0.12$ , respectively. The experimental data are reported in Figure 2 of the Supporting Information. The average  $R_1$ ,  $R_2$ , and NOE values for wild-type Cu<sub>2</sub>Zn<sub>2</sub>SOD are  $0.60 \pm 0.11 \text{ s}^{-1}$ ,  $33.9 \pm 4.1 \text{ s}^{-1}$ , and  $0.81 \pm 0.10$ , respectively (17).

The residues that exhibit conformational exchange processes from CPMG and  $R_{1\rho}$  analysis in G93A Cu<sub>2</sub>Zn<sub>2</sub>SOD are Val7, Asn19, Gln22, Lys30, Gly37, Phe45, His48, Cys57, Val81, Ala95, Leu117, Lys122, Thr137, and Gly147. Furthermore, residues Phe20, Asp52, and Ser107 have  $R_2$  values higher than the average, indicating the presence of exchange processes occurring in less than 50  $\mu\text{s}$ . In wild-type

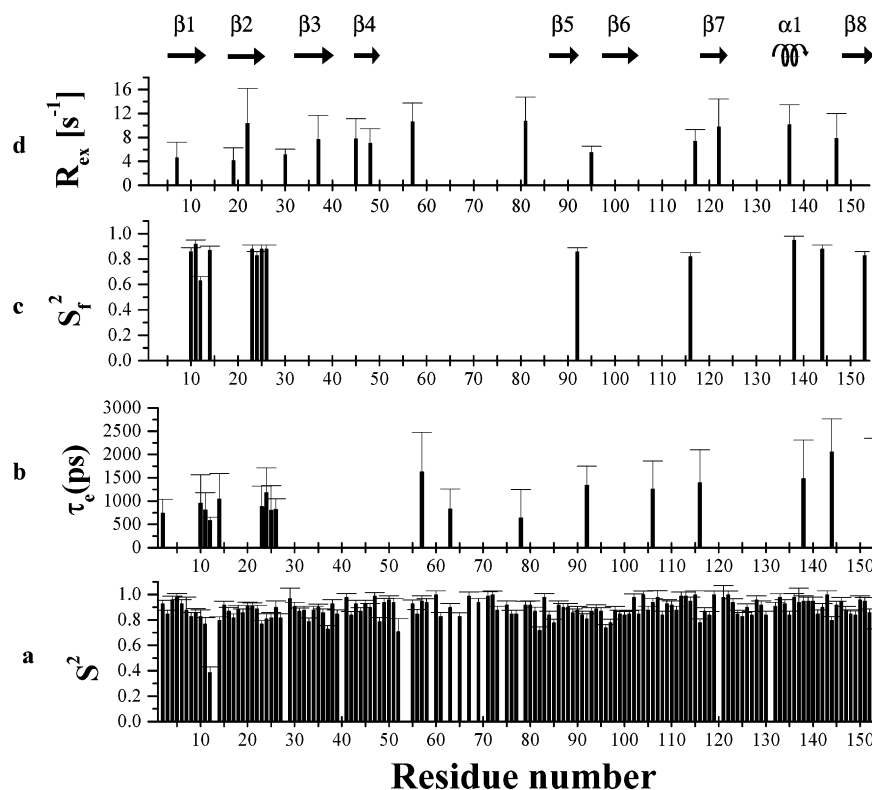


FIGURE 2: Model-Free parameters of G93A Cu<sub>2</sub>Zn<sub>2</sub>SOD. Shown are backbone amide order parameters  $S^2$  (a), the effective correlation time for these fast motions ( $\tau_e$ ) (b), the order parameters for the fast motions ( $S_f^2$ ) (c), and the conformational exchange contribution to  $R_2$  as determined from CPMG and  $R_{1\rho}$  analysis as a function of residue number. The secondary structure elements are reported at the top.

Cu<sub>2</sub>Zn<sub>2</sub>SOD, the residues that exhibit conformational exchange processes from  $R_{1\rho}$  analysis are Asn19, Phe20, Thr54, Cys57, Thr58, Ser107, Leu117, Lys122, Leu126, Ala145, and Ile151 (16). Residue Gly127 has an  $R_2$  value that is higher than the average.

**Analysis of the Relaxation Data.** Within the Model-Free analysis, the crystal structure of the G37R mutant of human Cu<sub>2</sub>Zn<sub>2</sub>SOD (16) was used to calculate the inertia tensor as was done for the wild-type Cu<sub>2</sub>Zn<sub>2</sub>SOD (17). The relaxation data are best fit by an axial diffusion tensor with the  $D_{||}/D_{\perp}$  ratio equal to  $1.26 \pm 0.06$ . The same value is also found if the solution structure of wild-type Cu<sub>2</sub>Zn<sub>2</sub>SOD (48) is used. For wild-type Cu<sub>2</sub>Zn<sub>2</sub>SOD, the  $D_{||}/D_{\perp}$  ratio was found to be  $1.24 \pm 0.02$ , which is essentially equal to that of the mutant described here. The  $\tau_m$  value obtained from the fitting of the relaxation data for G93A Cu<sub>2</sub>Zn<sub>2</sub>SOD is  $22.9 \pm 1.6$  ns which compares well with a value of  $25.3 \pm 2.1$  ns for wild-type Cu<sub>2</sub>Zn<sub>2</sub>SOD (17).

The average  $S^2$  value is  $0.89 \pm 0.04$  (Figure 2a) which is comparable to the average  $S^2$  value obtained for wild-type Cu<sub>2</sub>Zn<sub>2</sub>SOD of  $0.92 \pm 0.04$  (17). Most of the residues are fitted using  $S^2$  as the only variable parameter (model 1). Values for the effective internal correlation time  $\tau_e$  can be determined for 18 residues (Figure 2b), all located in the loop regions. Some of the latter residues are better described with two classes of internal motions (model 5), and the order parameter ( $S_f^2$ ) for the fast internal motions can be determined (Figure 2c). The latter residues are mainly located at the N-terminus of the protein and are all clustered in two loop regions experiencing collective motions, with some scattered residues throughout the length of the protein (residues 116 and 138).

Conformational exchange contributions to  $R_2$  from Model-Free analysis ( $R_{ex}$ ) were found for Phe20, Asp52, Ser107, and Thr137, consistent with their  $R_2$  values being higher than the average. The relaxation parameters of residues Thr54, Ser68, and Glu78 cannot be fitted within the Model-Free analysis.

## DISCUSSION

The structure of G93A Cu<sub>2</sub>Zn<sub>2</sub>SOD is very similar to that of wild-type Cu<sub>2</sub>Zn<sub>2</sub>SOD since the <sup>15</sup>N-HSQC spectra and NOE patterns of the two proteins are nearly identical with the exception of two distinct segments (residues 35–42 and 92–95) shown in Figure 1. These two areas showing <sup>15</sup>N and <sup>1</sup>H chemical shift differences between wild-type and G93A Cu<sub>2</sub>Zn<sub>2</sub>SOD involve loops III and V. The shift differences are appreciable enough to indicate local structural variations in these two regions of the protein. It is clear that the disruption of local interactions between these two conserved areas induced by the mutation alters the local structure as well as enhances the backbone mobility in these loops. The differences in chemical shifts found in these regions correlate well with the observation that electrostatic and hydrophobic interactions between these two regions significantly contribute to the stability of the  $\beta$ -barrel structure of Cu<sub>2</sub>Zn<sub>2</sub>SOD (11, 12, 16, 48). There is also a slight chemical shift difference for Gln15 in G93A Cu<sub>2</sub>Zn<sub>2</sub>SOD when compared to the wild type, indicating a structural rearrangement in the area near Val14 which stabilizes loop III by interacting with Leu38 (16). Moreover, His43 H $\epsilon_1$ , which is involved in hydrogen bonding to the carbonyl oxygen of residues 39 and 40 (48), is shifted considerably

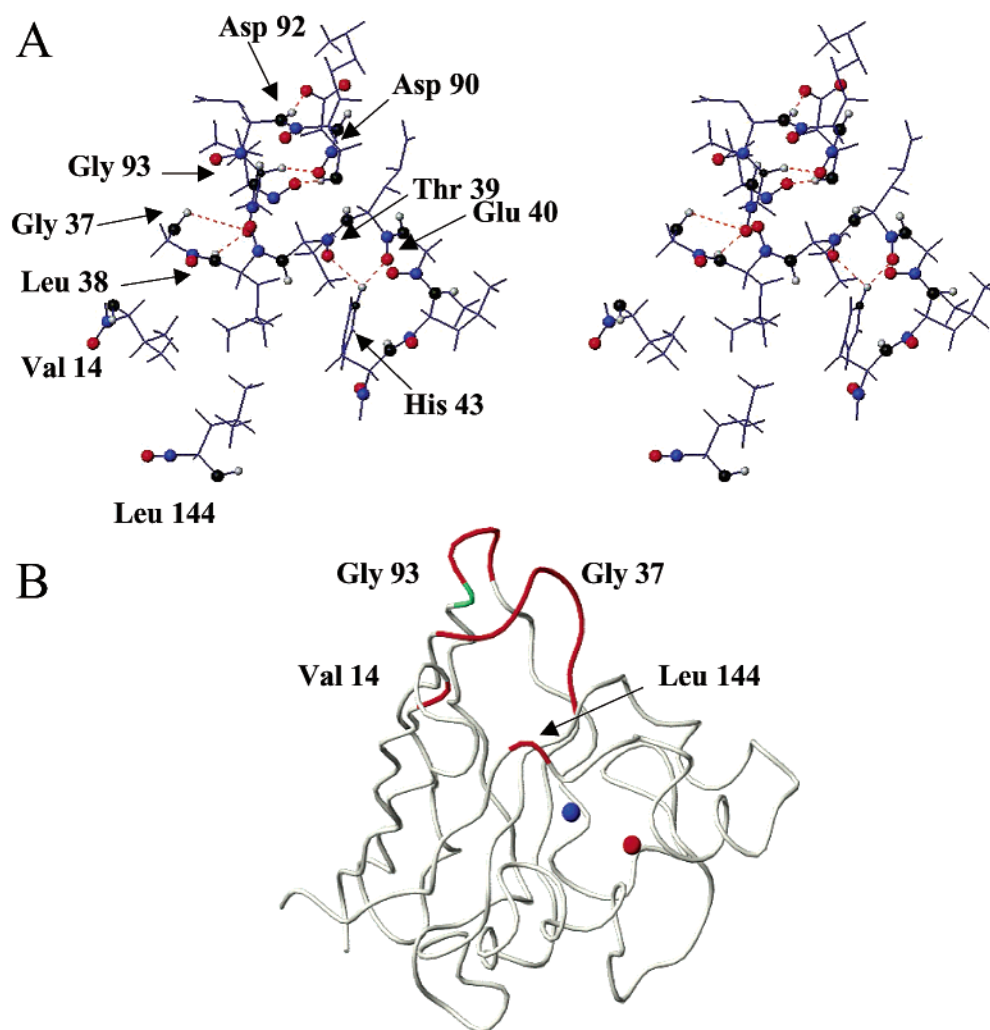


FIGURE 3: Hydrogen bond network in the  $\beta$ -barrel plug of  $\text{Cu}_2\text{Zn}_2\text{SOD}$ . The hydrogen bond network is taken from the solution structure of  $\text{Cu}_2\text{Zn}_2\text{SOD}$  (48). The residues that comprise the  $\beta$ -barrel plug (3, 12) are depicted in panel A. Carbon is shown in blue, oxygen in red, nitrogen in black, and hydrogen in gray. The hydrogen bonds are represented as orange dashed lines. (B) The ribbon diagram highlights the protein structure of the region shown in panel A. The green segment indicates Gly93.

in G93A  $\text{Cu}_2\text{Zn}_2\text{SOD}$  relative to the wild type (Figure 1, inset).

The average backbone  $S^2$  value for the G93A  $\text{Cu}_2\text{Zn}_2\text{SOD}$  mutant is  $0.89 \pm 0.04$ , indicating an essentially rigid protein.  $S^2$  values in this range are typically seen in proteins such as  $\text{Cu}_2\text{Zn}_2\text{SOD}$  that contain a large amount of  $\beta$ -strands in their secondary structures (17, 49, 50). Most of the residues of G93A  $\text{Cu}_2\text{Zn}_2\text{SOD}$  located in the  $\beta$ -strands, with the exception of strand  $\beta_6$  which follows the G93A mutation (residues 94–101), have high  $S^2$  values indicating high rigidity of the protein (Figure 2a).

The G93A mutation occurs in loop V of  $\text{Cu}_2\text{Zn}_2\text{SOD}$  and is in an area of the protein in which hydrogen bonding interactions are thought to facilitate the “plug” end of the  $\beta$ -barrel (3, 48). In addition, there is evidence to suggest that the presence of the left-handed Gly93 residue is critical in maintaining packing interactions in this area of the protein (51). Therefore, it is not surprising that residues around the mutation experience sizable internal mobility. Strand  $\beta_6$  is also significantly more mobile in the mutant than in wild-type  $\text{Cu}_2\text{Zn}_2\text{SOD}$  (average  $S^2$  of  $0.93 \pm 0.04$  in the wild type vs  $0.84 \pm 0.03$  in the mutant). This may be also a result of the disruption of the hydrogen bond network in loop V due

to the mutation. The hydrogen bond network around Gly93 found in the solution structure of human  $\text{Cu}_2\text{Zn}_2\text{SOD}$  is shown in Figure 3 (48). Hydrogen bonds between the Asp92–Asp90, Asp90–Val94, and Gly93–Asp90 pairs aid in stabilizing the turn in loop V (11, 16, 48, 52). This decrease in rigidity may play a role in the in vivo instability previously found in G93C, another ALS mutant with a substitution at residue 93 (6).

Contrary to the behavior of  $\beta$ -strands, residues located in loops are characterized by  $S^2$  values that are lower than the average (Figure 2a). Particularly striking is the very low  $S^2$  value of Gly12 ( $0.39 \pm 0.04$ ) which is located in loop I. Interestingly, the residues around the electrostatic channel (loop VII) in G93A  $\text{Cu}_2\text{Zn}_2\text{SOD}$  do not exhibit internal mobility. This area is instead characterized by large  $S^2$  values.

Values for the internal correlation internal time ( $\tau_c$ ) and fast order parameter ( $S_f^2$ ) are found for residues clustered in two loop regions, and these regions experience collective motions (Figure 2b,c). This is expected since loop regions normally display greater mobility than residues involved in secondary structural elements. Conformational exchange processes are involving residues scattered along the protein

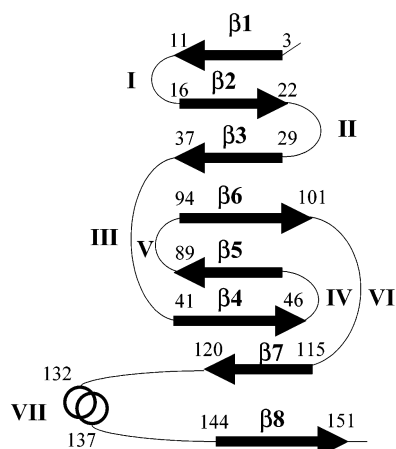


FIGURE 4: Cu<sub>2</sub>Zn<sub>2</sub>SOD secondary structure.  $\beta$ -strands, as found in the solution structure of wild-type Cu<sub>2</sub>Zn<sub>2</sub>SOD (48), are numbered beginning from the N-terminus of the protein and loop regions are designated by Roman numerals.

backbone that are present both in the long loop IV and in several  $\beta$ -strands.

**Comparison of the Backbone Dynamics and NMR Parameters of G93A and Wild-Type Cu<sub>2</sub>Zn<sub>2</sub>SOD.** To compare the backbone mobilities of the G93A Cu<sub>2</sub>Zn<sub>2</sub>SOD mutant and wild-type Cu<sub>2</sub>Zn<sub>2</sub>SOD, the differences in  $S^2$  values for both proteins are displayed in Figure 5. A positive value indicates that the H–N vector is more rigid in wild-type Cu<sub>2</sub>Zn<sub>2</sub>SOD than in G93A Cu<sub>2</sub>Zn<sub>2</sub>SOD, whereas a negative value indicates the opposite. Interestingly, the G93A Cu<sub>2</sub>Zn<sub>2</sub>SOD protein appears to be more mobile than the wild type, which is characterized by an overall average  $S^2$  value of  $0.92 \pm 0.04$ , at many residues located throughout the protein. The most notable differences in  $S^2$  are observed for residues 10–17, 35–37, 76–84, 92–103, 117, 127, 144, 148, and the final two C-terminal residues, which are depicted in Figure 6. Interestingly, the residues in the electrostatic channel (loop VII) do not experience an appreciable difference in mobility relative to the wild type. This result seemingly rules out the possibility that a large molecule binds nonspecifically to the copper site as a consequence of increased flexibility around the electrostatic channel. However, the increased overall dynamics of the mutant protein may still provide increased access of small molecules, such as H<sub>2</sub>O<sub>2</sub> or peroxynitrite, to the copper site, resulting in potentially toxic free radical chemistry (53, 54).

The packing interactions found between Val14, residues 35–38, and Leu144 also contribute to the formation of the  $\beta$ -barrel plug (3, 48) (Figure 3). These residues were also found to exhibit more mobility than the wild type. These findings strengthen the idea that the substitution of glycine with alanine at position 93 disrupts the normal hydrogen bond network and renders G93A Cu<sub>2</sub>Zn<sub>2</sub>SOD more mobile than the wild type in a global as well as local sense.

Another region of G93A Cu<sub>2</sub>Zn<sub>2</sub>SOD that has  $S^2$  values significantly lower than those reported for the wild-type protein is the segment between residues 76 and 84, with average  $S^2$  values of  $0.86 \pm 0.03$  in the mutant and  $0.95 \pm 0.03$  in the wild type. These residues comprise roughly half of the zinc subloop and contain two of the zinc binding ligands. This enhanced flexibility at the zinc site is not due to the absence of zinc since the G93A Cu<sub>2</sub>Zn<sub>2</sub>SOD sample

studied here was found to be fully metalated with two copper and two zinc ions per protein dimer (see Experimental Procedures). Moreover, to our knowledge, there is no evidence that indicates a zinc deficiency *in vivo* in the majority of FALS mutant Cu<sub>2</sub>Zn<sub>2</sub>SOD proteins (14, 55). Nevertheless, previous studies on the zinc binding properties of isolated Cu<sub>2</sub>Zn<sub>2</sub>SOD-related FALS mutants do suggest that the zinc sites of these proteins are altered in some fashion (56, 57). The enhanced mobility in this region may be related to these altered binding properties of the zinc site in the FALS mutant proteins.

Surprisingly, there are a number of residues in the wild-type protein with  $S^2$  values lower than those observed in G93A Cu<sub>2</sub>Zn<sub>2</sub>SOD (Figure 5). The residues also display conformational exchange processes in the wild type with the exception of Gly108 (17), whereas they do not in the mutant. Two of these residues, Ala89 and Val94 which are spatially close to each other, may be stabilized in the mutant via hydrophobic interactions. Since Asp90, Lys91, Asp92, and Ala93 are more flexible due to the mutation, it is reasonable to assume that Ala89 and Val94 might be closer in space and thereby positioned within the van der Waals radius of interaction. Perhaps long-range effects induced by the increased mobility around the  $\beta$ -barrel plug (3) create steric constraints around these residues, consequently rendering them less mobile.

The differences in mobility (in terms of  $S^2$ ) between two species are related to their differences in entropy ( $\Delta S$ ) via eq 10. The validity of this equation is limited to residues with  $S^2$  values of  $<0.95$  (46). If residues with a value of  $S^2$  higher than this limit are excluded, the entropy difference between G93A and wild-type Cu<sub>2</sub>Zn<sub>2</sub>SOD is estimated to be  $40 \pm 13 \text{ J mol}^{-1} \text{ K}^{-1}$ . This value indicates a global higher disorder in G93A Cu<sub>2</sub>Zn<sub>2</sub>SOD. However, entropy variations calculated from order parameters should not be taken as the total entropic difference since the former contribution reflects only differences in rapid motions detected by NMR. Of course, additional factors are expected to contribute to the total entropy difference.

The medium- and long-range NOEs observed in the wild-type protein between residues in loops I, III, and V and in strand  $\beta_6$  are either not present or dramatically decreased in intensity in G93A Cu<sub>2</sub>Zn<sub>2</sub>SOD. This observation directly correlates with the increased internal mobility in the mutant.

**Biological Implications of the G93A Mutation.** Mice expressing Cu<sub>2</sub>Zn<sub>2</sub>SOD mutants have been seen to contain intracellular plaques or inclusions in their motor neurons, most of which occur before the onset of the phenotypic observations associated with ALS (58). In particular, spinal cords of mice expressing G93A were found to contain insoluble high-molecular weight complexes composed of G93A (59). These data suggest a tendency of Cu<sub>2</sub>Zn<sub>2</sub>SOD mutants to aggregate due to the decreased stability of the protein. Although G93A was found to be similar in terms of thermal stability to the wild type (data not shown), over time it shows a slow denaturation which might be linked to the increased mobility in many areas of the protein. Moreover, recent structural analysis suggests that  $\beta$ -barrel folds have continuous hydrogen bond networks around the ends of the  $\beta$ -cylinder to protect proteins from aggregation arising from edge-to-edge contacts between neighboring  $\beta$ -sheet proteins (60). The loss of hydrogen bonding around



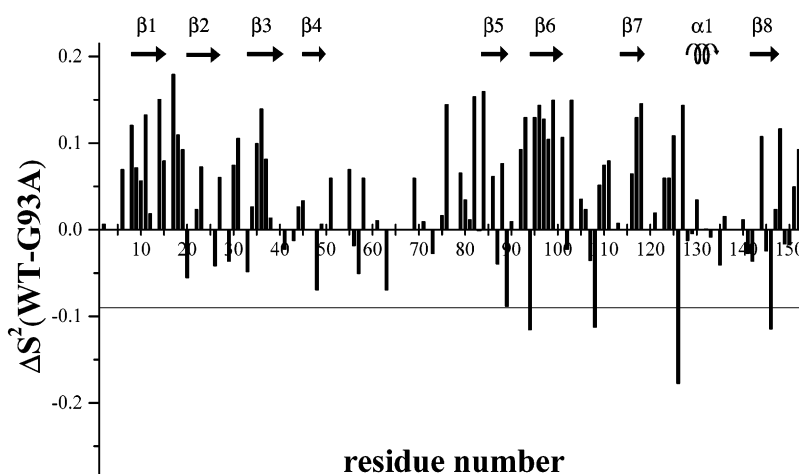


FIGURE 5: Differences in order parameters between wild-type and G93A Cu<sub>2</sub>Zn<sub>2</sub>SOD. A positive value indicates that the H–N vector is more rigid in wild-type Cu<sub>2</sub>Zn<sub>2</sub>SOD than in G93A Cu<sub>2</sub>Zn<sub>2</sub>SOD, whereas a negative value indicates the opposite. The two thin lines represent the sum of the average errors from fitting. The secondary structure elements are reported at the top.

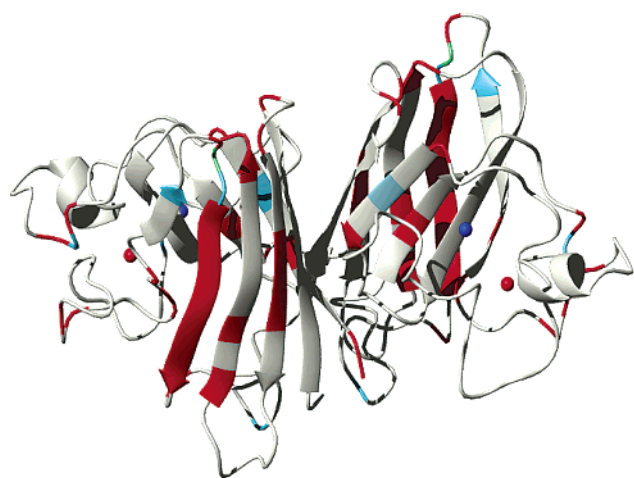


FIGURE 6: Differences in internal mobility between wild-type and G93A Cu<sub>2</sub>Zn<sub>2</sub>SOD in  $S^2$ . The solution structure of Cu<sub>2</sub>Zn<sub>2</sub>SOD (48) was used to illustrate the areas with markedly different  $S^2$  values between the two systems. Only residues with  $S^2$  values differing by more than  $\pm 0.09$  (sum of the error) are colored. Red depicts residues in G93A Cu<sub>2</sub>Zn<sub>2</sub>SOD with a lower and blue with a higher order parameter than wild-type Cu<sub>2</sub>Zn<sub>2</sub>SOD. Green indicates the position of the G93A mutation.

the  $\beta$ -barrel plug as well as the increase in NH mobility, observed in G93A, might favor edge-to-edge aggregation and  $\beta$ -sheet amyloid fibrillogenesis (60).

The large number of Cu<sub>2</sub>Zn<sub>2</sub>SOD-related FALS mutations at residue 93 indicates that the loss of the glycine and not the addition of an alanine is an important factor in the pathology arising from G93A Cu<sub>2</sub>Zn<sub>2</sub>SOD. The loss of rotational degrees of freedom brought about by replacing the glycine increases the mobility of the residues surrounding the mutation. This resulting increase in mobility may create a dynamic transient opening in the  $\beta$ -barrel plug of G93A Cu<sub>2</sub>Zn<sub>2</sub>SOD that contains dissimilar electrostatic characteristics with respect to the active site channel, subsequently allowing small molecules to bind the protein and possibly facilitating G93A aggregation. Either a combination of edge-to-edge interactions, aberrant ligand binding, or both may explain the motor neuron-specific aggregation of Cu<sub>2</sub>Zn<sub>2</sub>SOD observed in FALS transgenic mice (59). The enhanced mobility of strand  $\beta$ 6 immediately following the G93A

Cu<sub>2</sub>Zn<sub>2</sub>SOD mutation may also aid in the creation of such a binding site. It is conceivable that the creation of an aberrant binding site in G93A Cu<sub>2</sub>Zn<sub>2</sub>SOD may aid in bringing about the toxic properties associated with Cu<sub>2</sub>Zn<sub>2</sub>SOD FALS mutants.

In this study, we have found that G93A Cu<sub>2</sub>Zn<sub>2</sub>SOD displays enhanced backbone mobility relative to wild-type Cu<sub>2</sub>Zn<sub>2</sub>SOD on the nanosecond to picosecond time scale stemming from the loss of important interactions between residues that comprise the  $\beta$ -barrel plug (3, 48). Chemical shift analysis and NOE analysis have also revealed structural differences between the residues involved in this region. There are also a number of other Cu<sub>2</sub>Zn<sub>2</sub>SOD-related FALS mutations located in this critical area such as G37R, L38V, L38R, H43R, A89V, D90A, and D90V (up to date list of ALS mutations available at [www.alsod.org](http://www.alsod.org)). The G37R Cu<sub>2</sub>Zn<sub>2</sub>SOD mutant is also thought to be flexible in this region as indicated by the presence of high atomic displacement parameters in the crystal structure (16). Because of the high level of conserved residues in this region and the number of FALS-related Cu<sub>2</sub>Zn<sub>2</sub>SOD mutants found within, it is readily apparent that the disruption of this critical region of the protein plays a significant role in the creation of the pathological properties associated with Cu<sub>2</sub>Zn<sub>2</sub>SOD mutants.

The pathological mechanism of FALS-associated Cu<sub>2</sub>Zn<sub>2</sub>SOD mutants remains an elusive target. However, it is becoming clearer that discrete structural instability and dynamic properties of these mutants are likely to play an essential role in the etiology of the disease. In this study, we have presented convincing evidence that G93A Cu<sub>2</sub>Zn<sub>2</sub>SOD displays dynamic and structural characteristics not present in the wild-type protein, which undoubtedly contribute to the pathology of FALS-related Cu<sub>2</sub>Zn<sub>2</sub>SOD. Further studies are being carried out to gain more insight into how these distinct characteristics of G93A Cu<sub>2</sub>Zn<sub>2</sub>SOD selectively affect motor neurons at advanced stages of life.

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## SUPPORTING INFORMATION AVAILABLE

Examples of  $^{15}\text{N}$   $R_1$  and  $R_2$  relaxation data (Figure 1) and  $R_1$ ,  $R_2$ , and NOE values of G93A Cu<sub>2</sub>Zn<sub>2</sub>SOD (Figure 2). This material is available free of charge via the Internet at <http://pubs.acs.org>.

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